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11) Which molecule has a dipole moment of zero? A) CHCl_3 B) CH_2Cl_2 C) $\text{ClHC}=\text{CH}_2$ D) trans- $\text{ClHC}=\text{CHCl}$ E) None of these choices. Answer: D Topic: Molecular geometry, polarity Section Reference 1: 1.5 and 2.3 Difficulty: Medium 12) Which molecule would have a dipole moment greater than zero? A) BeCl_2 B) BCl_3 C) CO_2 D) H_2O E) CCl_4 Answer: D

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Find an easier way to learn organic chemistry with *Arrow-Pushing in Organic Chemistry: An Easy Approach to Understanding Reaction Mechanisms*, a book that uses the arrow-pushing strategy to reduce this notoriously challenging topic to the study of interactions between organic acids and bases. Understand the fundamental reaction mechanisms relevant to organic chemistry, beginning with S_N2 reactions and progressing to S_N1 reactions and other reaction types. The problem sets in this book, an excellent supplemental text, emphasize the important aspects of each chapter and will reinforce the key ideas without requiring memorization.

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